



Impact of sorbent injection on ash behavior in the small and pilot scale

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BiOxySorb Final Workshop Stuttgart, Germany 22-23 November, 2016



Economic low carbon power Production and Emissions Control for Future and Flexible Biomass Co-fired Power Stations

Outline

- \circ Objective
- Background
- \circ Motivation
- Methodology
- \circ Results
- \circ Conclusion





Objective: Acid gas removal



- Why limit acid gas (SOx and HCl) emissions
 - Cause harmful impacts to the environment and human health
 - High and low temperature corrosion
- Approaches to limiting acid gases
 - Limiting the input of S and Cl with the fuel input
 - Using different scrubbing technologies for acid gas removal
 - Dry sorbent injection (DSI) systems
 - Spray drying absorption (SDA) processes (semi-dry flue gas cleaning process)
 - Wet flue gas desulfurization processes
 - Circulating fluidized bed (CFB) scrubbers







Motivation: Fly ash major applications



Table : energy demand for production

	Total Energy Required kJ/kg					
Material	of Material					
Cement	372					
Lime	173					
Fly ash	0					

(Source: (Lohtia 1995))

Use of fly ash in the construction industry and underground mining in Europe for year 2010, (EU 15)



Source: (Feuerborn 2015)



Motivation: Fly ash use benefits





Source: (modified from Sankaralingam 2012)

Resource conservation



Motivation: Concerns of dry sorbent injection



- Potential fly ash concerns in normal concrete from calcium based sorbent injections
 - 1. Increased sulfur content in fly ash

Sulfate reactions create compounds occupying a greater volume, causing expansion and failure of concrete

- Increased free calcium content in fly ash Excess amounts react with water forming Ca(OH)₂, undesirable volume changes in concrete
- Increased chloride content in fly ash
 Cl ions negatively impact the protective film covering steel reinforcement in concrete (corrosion risk)





Methodology





• Oxy-fuel condition with sorbent in-furnace injection



Test facility: 20kW lab-scale

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stack carrier gas air ₩₩₩ _____П primary gas 🖂 0₂/C0₂ burner secondary gas Air and oxy-fuel NOx baghouse SO_2 filter conditions Hg^₀ H_2O ž insulation 0-In-furnace and before O. filter sorbent electrical heating 0-0- $\frac{1}{2}$ injection under air gas sampling \odot burnout gas 0ports 0combustion sorbent injec-**(+)** tion locations 0gas sampling probe 3 7777777 multiple zone electrically heated flue gas duct

fuel feed

Bioxy Sorb

Fuel and sorbent characteristics



> Coal analysis

Fuel	NCV	H ₂ O	Ash	V	C _{fix}	C	S	H	N	O
	[MJ/kg, raw]	[%, raw]	[%, wf]	[%, waf]	[%, waf]	[%, waf]	[%, waf]	[%, waf]	[%, waf]	[%, waf]
US2.5 coal	30.5	1.6	9.43	38.72	61.28	81.91	2.67	5.20	1.59	8.63

Fuel	AI2O3	CaO	Fe2O3	к2О	MgO	Na2O	P2O5	SO3	SiO2	TiO2
US2.5 coal	20.3	3.7	14.4	2.4	0.9	0.7	0.2	3.0	52	1.1

Sorbent analysis
Sorbacal®SPS mainly Ca(OH)₂.



Methodology: Fly ash characterizations



\circ Chemical composition

- Oxides analysis by ICP-OES
- Reactive SiO₂
- Reactive CaO

o Mineralogical classification

X-ray powder diffraction (XRD)

Classification	Mineral phases
Active	Anhydrate, Lime
Semi-active	Carbonates, Calcite, Hematite
Inert	Quartz, Mulite
Pozzolanic	Glass (amorphous material)

O Physical charateristics

Particle size ditribution (PSD), Malvern 2600



Methodology: Fly ash utilization



European Union standards

- Fly ash for concrete Part 1: Definition, specifications, and conformity criteria (EN 450-1) *(high technical application)*
- Hydraulic road binders- Part 1: Rapid hardening hydraulic road binders- composition, specifications, and conformity criteria (EN 13282-1)
- Hydraulic road binders- Part 2: Normal hardening hydraulic road binders- composition, specifications, and conformity criteria (EN 13282-2)
- Cement- Composition, specifications and conformity criteria for common cements (EN 197-1)



Results: Fly ash characteristics



Chemical composition (500 kW)





- o Increased S content of the ash with sorbent injection
- Capture of sulfur with finer particles finding in ESP2 and ESP3



Fly ash enrichment of elements (500 kW)



	Dilution factor
Al ₂ O ₃	1.5
Fe ₂ O ₃	2.2
MnO ₂	1.3
SiO ₂	2.3
Average	1.9

Dilution factor =
$$D_A = \frac{X_{(i,A)}}{X_{(i,A+S)}}$$

 $X_{(i,A)}$: Oxide concentration in the fuel ash $X_{(i,A+S)}$: Oxide concentraion in the fly ash

Ca and S mass based enrichment factor

Са	12
S	14

Enrichment factor
$$= \frac{X_{(i,A+S)}}{X_{(i,A)}} \times D_A$$



XRD analysis (500 kW)



Oxyfuel scenario

Oxy-fuel with SPS in-furnace injection



• Reduced amorphous material and increased active phase with sorbent addition



XRD analysis (500 kW)



Scenario	Quartz	Anhydrite	Lime	Hematite	Mullite	Amorphous matter
Oxyfuel	+++	++	-	++	++	+++
Oxyfuel with SPS	++	+++	++	+	+	-

+++ significant; ++ moderate; + insignificant; - none

Scenario	Effect
Oxyfuel	Pozzolanic quality fly ash- react with Ca(OH)2 to produce C-S-H
Oxyfuel with SPS	Shifted to mineral phases that have an active or semi-active quality



Fly ash particle size distribution (500 kW)





- Reduced particle size distribution due to fine sorbent particles
- Smaller particles sizes and higher surface area leading to more reactivity
 → improve fly ash performance in concrete



Chemical composition (Air and oxy-fuel 20 kW)



- No difference in chemical composition under air and oxy-fuel conditions
- Increased S content for in-furnace injection
- \circ $\,$ Increased CI content for before filter injection $\,$



XRD analysis (Air and Oxy-fuel, 20 kW)





No significant difference between the mineral faces in air and oxy-fuel combustion conditions



XRD analysis (Sorbent injection, 20 kW)





• Reduced amorphous background and more active material CaSO₄ formation



XRD anlysis (Sorbent injection, 20 kW)





• Reduced amorphous background and more unreacted Ca(OH)2





Results: Fly ash marketability



Fly ash marketability for concrete (EN450-1)



EN450-1 (for concrete) Some parameters	Value [% mass]	Oxyfuel (500kW)	Oxyfuel, SPS in- furnace (500 kW)	Air (20 kW)	Oxy-fuel (20kW)	Air, SPS in- furnace (20 kW)	Air, SPS before filter (20 kW)
Loss on ignition class A class B class C	≤5 ≤7 ≤9	5.4	2.2 (4.18)	4	7	0.8	1.1
SiO ₂ -Al ₂ O ₃ -Fe ₂ O ₃	≥ 70	74.83	42.09	86	85	41.2	61.1
Reactive (CaO)	≤ 10	satisfied	9	10	10	15.9	17.7
Reactive SiO ₂	≥ 25	42	8.67	-	-	-	-
SO ₃	≤ 3	2.97	23.1	2.7	3.5	23.2	7.7
Chloride (Cl ⁻)	≤ 0.1	0.016	0.088	0.08	0.06	0.5	1.8

(Source: Feuerborn 2011)

- Oxy-fuel combustion does not have impact on fly ash utilization
- $\circ~$ Sorbent injection deteriorates the fly ash characteristic for EN450-1



Fly ash marketability for cement (EN 197-1)



Cement- Composition, specifications (EN 197-1)

EN 197-1	Value [% mass]	Oxyfuel (500kW)	Oxyfuel, SPS in- furnace (500 kW)	Air (20 kW)	Oxy-fuel (20kW)	Air, SPS in- furnace (20 kW)	Air, SPS before filter (20 kW)
Reactive (CaO)	≤ 10 (a) ≤ 10-15 (b)	satisfied	9	10	10	15.9	17.7
Reactive SiO2	≥ 25	42	8.67	-	-	-	-
SO ₃	≤ 4	2.97	23.1	2.7	3.5	23.2	7.7
Chloride (Cl ⁻)	≤ 0.1	0.016	0.088	0.08	0.06	0.5	1.8

(a) Siliceous fly ash : Reactive SiO₂ \ge 25% and reactive CaO <10%

(b) Calcareous fly ash : Reactive SiO2 \geq 25% and reactive CaO>10-15%

- LOI requirement is already satisfied in EN 450-1
- Fly ash obtained from sorbent injection do not meet the requirements for cement standards (EN 197-1)



Fly ash marketability for road binders (EN 13282)



Hydraulic road binders- (EN- 13282-1)

EN 13282-1	Value [% mass]	Oxyfuel (500kW)	Oxyfuel, SPS in- furnace (500 kW)	Air (20 kW)	Oxy-fuel (20kW)	Air, SPS in- furnace (20 kW)	Air, SPS before filter (20 kW)
Reactive SiO ₂	≥ 25	42	8.67	-	-	-	-
SO ₃	≤ 5 (a) ≤ 10 (b)	2.97	23.1	2.7	3.5	23.2	7.7

(a) Siliceous fly ash : Reactive SiO₂ \ge 25% and reactive CaO <10%

(b) Calcareous fly ash : Reactive SiO2 \ge 25% and reactive caO>10-15%

 Considering calcareous fly ash, SPS injection before the filter under air condition at 20 kW scale meets the requirement for road binders (EN 13282)



Fly ash marketability based on current evaluations and widely used standards





Potential innovative fly ash uses



• Innovative soil amendment

- Improve soil structure and physical characteristics
- Acid neutralizing reactions (calcium): releases nutrients; reduces plant toxicity (Al, Mn, Fe) in acidic soils
- SiO₂, CaO, Al₂O₃, SO₄ react to form secondary minerals: decrease toxic heavy metals leaching; improve water retention, soil strength, and porosity in soils (calcium alumina-silicate hydrates) Ca(OH)₂ → Ca²⁺ + 2OH⁻
 Ca²⁺ + 2OH⁻ + SiO₂ (clay silica) → CaO·SiO₂·H₂O
 Ca²⁺ + 2OH⁻ + Al₂O₃ (clay alumina) → CaO·Al₂O₃·H₂O Source: (Dermatas 2003)

• Innovative fertilizer

- Calcium oxide reaction with acidic components in soil improve plant fertility
- Ca and S are secondary plant nutrients

• Masonry constituent

- Aerated concrete bricks (principal constituents: quartz sand, pulverized fly ash, anhydrous calcium sulfate (CaSO₄)
- External render base coat component
 - High SO₃ content (acids resist weather and acid rain)
- Floor screeds



Conclusion



- Oxy-fuel combustion does not affect mineral phases of the fly ash
- Sorbent injection reduces the amourphases background of the fly ash and increases the active and semi active mineral phases
- Oxy-fuel combustion generates a fly ash quality applicable to normal coal ash standards (EN 450-1)
- Before the filter injection of SPS generates a fly ash applicable for hydraulic road binders from chemical analyses of limits
- In-furnace injection of SPS generates a fly ash quality with potential benefits in innovative land use, masonry or mortar, and etc. (i.e. take advantage of the behavior of the calcium minerals phases)







Thank you for your attention!

ACKNOWLEDGMENTS

The research leading to these results has received funding from the European Union's Research Fund for Coal and Steel (RFCS) research programme under grant agreement num. RFCR-CT-2013-00010 (RFCS research project BiOxySorb: <u>http://bioxysorb.eu-projects.de/</u>). The authors gratefully acknowledge this financial contribution and the support by advices and expertise of the BiOxySorb project partners Fundación Ciudad de la Energía, Uniper Technologies Limited, Lhoist Recherche et Développement SA and Gestamp Biomass Solutions. The authors also thank all colleagues of IFK's department "Firing Systems" who contributed to this work and in particular S. Pek, B. Ebner, M. Faulhaber, M. Pagano and T. Wagner, as well as W. Ross and his team of IFK's 'Laboratory for Fuels, Ashes and Slag' for their support in the performed experiments.

