



# Dry powdered lime for acidic gas removal in coal fired power plant

Impact on the fly ash behavior and valorization

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**Who is Lhoist ?**

**Lime based dry sorbent injection at 20MW pilot unit in CIUDEN**

**Dry sorbents delivered to CIUDEN**

**SO<sub>2</sub> removal performance in CIUDEN**

**Fly ash analysis & valorization**

**Conclusions**





# The Lhoist Philosophy

## Who are we ? What are we?

### ■ Our Philosophy



Quarry in Belgium  
Lhoist Group Collection



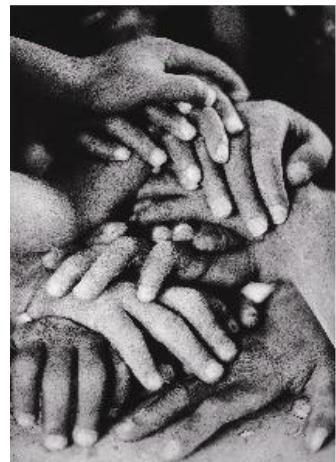
The Soldier's Tree, 1993  
Rodney Graham  
Lhoist Group Collection



Green Ball, 1995  
Gabriel Orozco  
Lhoist Group Collection



Untitled (Cowboys),  
1990  
Richard Prince  
Lhoist Group Collection



Les joueurs, 1991  
Michel François  
Lhoist Group Collection

**WE ARE PROUD  
INDUSTRIALISTS**

**WE ARE A  
FAMILY-OWNED  
COMPANY**

**WE ARE  
DRIVEN BY OUR  
CUSTOMERS AND  
MARKETS**

**WE ARE  
PIONEERS**

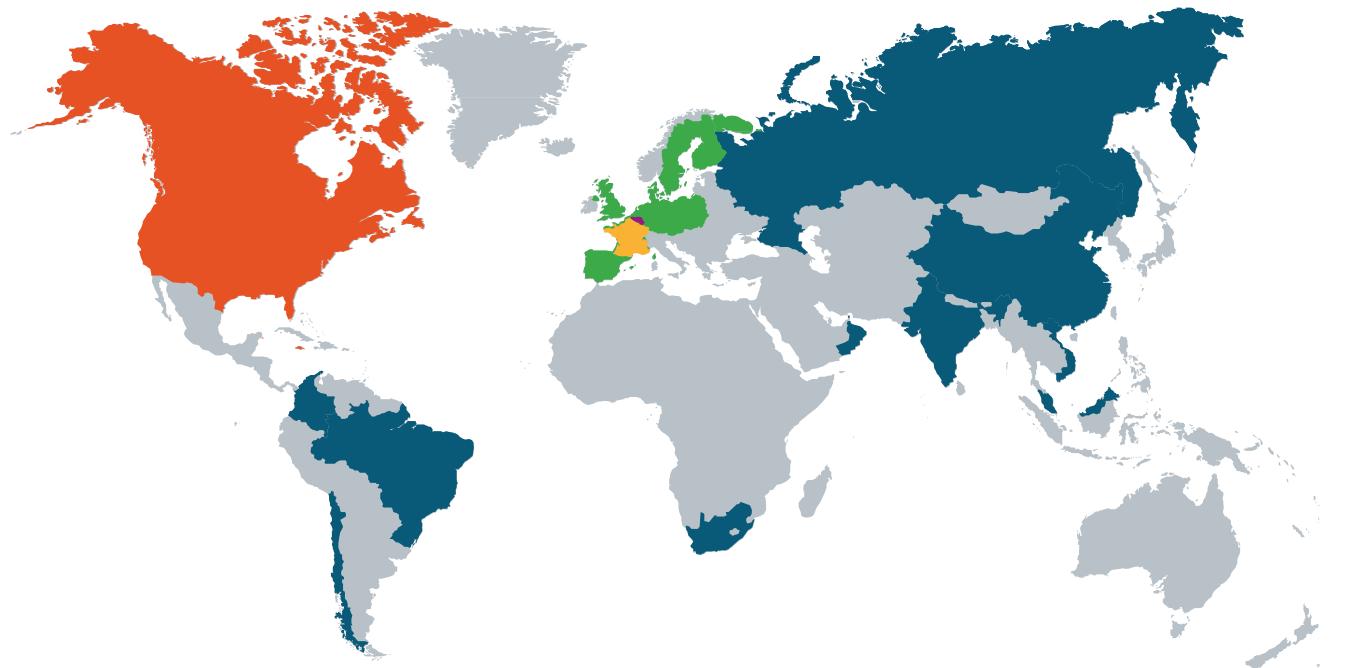
**OUR PEOPLE  
MAKE IT ALL  
HAPPEN**



# Lhoist Worldwide Presence

## Where are we?

- **Lhoist Group in the world**



**OVER 125 YEARS OF EXPERTISE & INNOVATION**

**1889**

Foundation of  
the Group in  
Belgium

**1926**

First expansion  
to France

**The 80's**

Crossing the  
Atlantic (US,  
Canada)

**The 90's**

Further  
development to  
Western &  
Eastern Europe

**New millennium**

South America,  
Africa, Asia  
& Russia



# Lhoist Group Key Numbers

## What is our size?

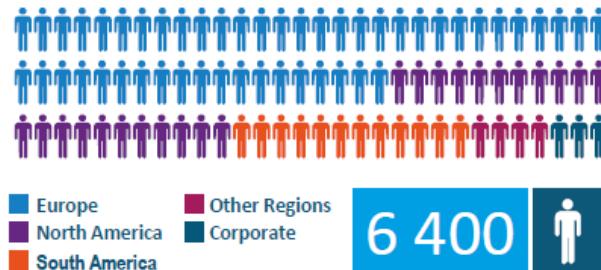
### ■ Lhoist Group



TURNOVER



>40 NATIONALITIES



- 1 R&D CENTER
- 4 CUSTOMER TECHNICAL SERVICE CENTERS
- > 25 COUNTRIES
- 100 PLANTS
- 800 FGT CUSTOMERS in EU and > 250 in the USA

# Lhoist Group Main Products

## What do we manufacture?



### MINERALS

- Limestone
- Dolomite
- Clay and others

### CALCINED PRODUCTS

- **Quicklime**  
(pebble, ground, milled)
- **Burnt dolomite**  
(soft-/hard-/overburnt)
- **Hydrated lime**

### SPECIALTY PRODUCTS

- Milk of lime
- Fluidized lime
- Low dust emission lime
- High porosity hydrated lime
- Formulated products

$\text{CaCO}_3$

$\text{MgCO}_3 \cdot \text{CaCO}_3$

$\text{Ca}_x \text{Mg}_y \text{Al}_z \text{Si}_p \text{O}_q (\text{OH})_r$

$\text{CaO}$

$\text{MgO} \cdot \text{CaCO}_3$

$\text{MgO} \cdot \text{CaO}$

$\text{Ca}(\text{OH})_2, \text{Mg}(\text{OH})_2 \cdot \text{CaCO}_3$

$\text{Mg}(\text{OH})_2 \cdot \text{Ca}(\text{OH})_2$

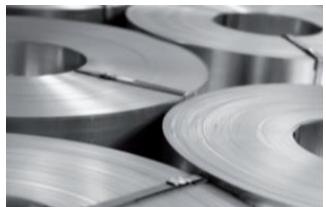
Formulated hydrates, clays, minerals & calcined products



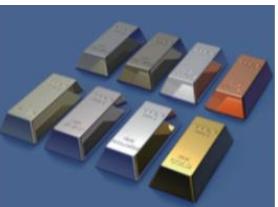
# Lhoist Group Main Markets

## Who are our customers?

### ■ Large diversity of applications



Iron and steel



Non-ferrous metals



Pulp and paper



Drinking water



Water and sludge



Flue gas treatment



Glass

Civil engineering

Building



Agriculture



Chemicals



Biofuels



## What is our market?

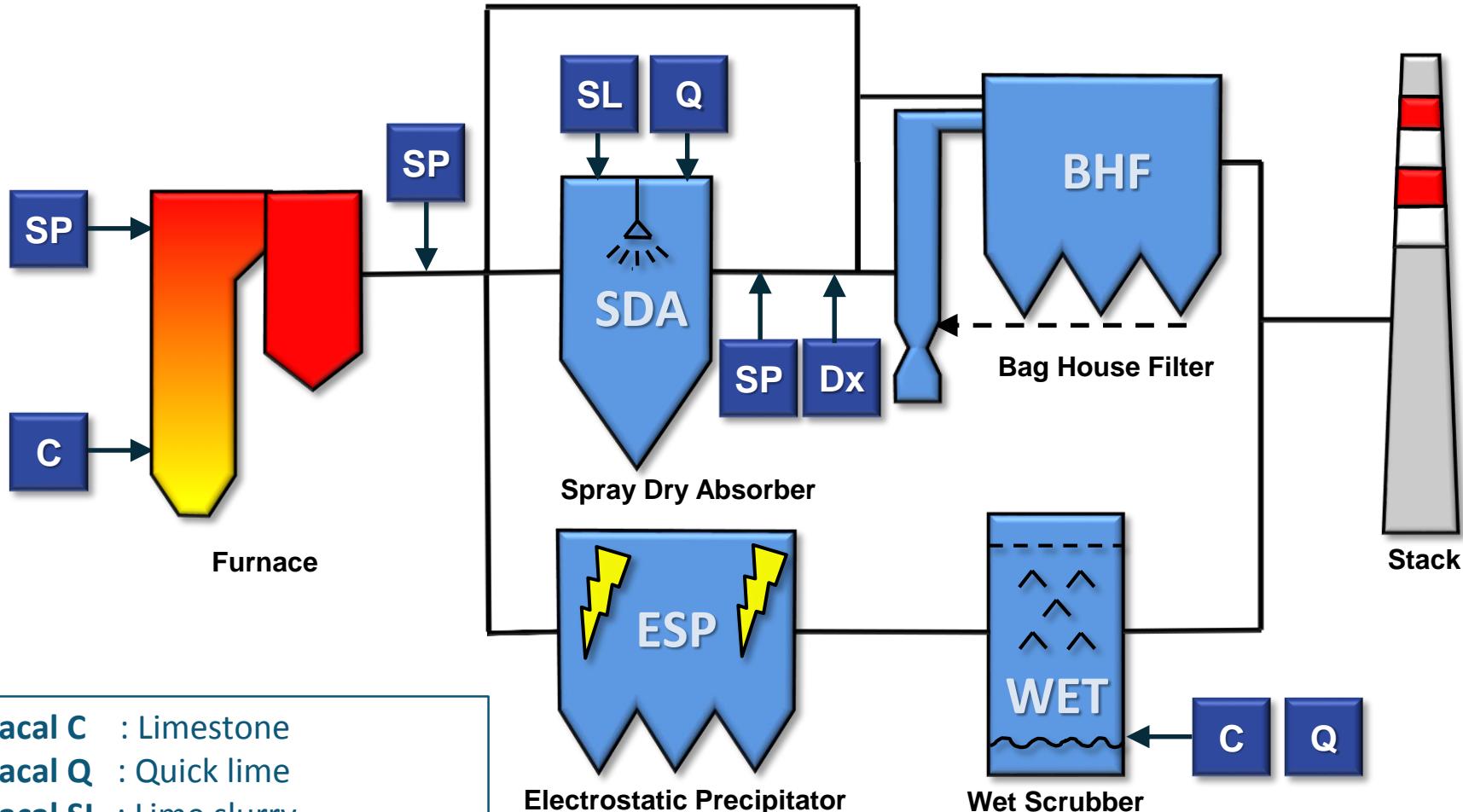
### ■ Turnover by Market

- Environmental applications are important and growing

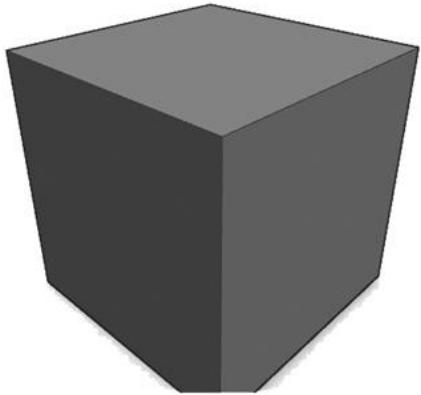


# Flue Gas Treatment with Calcium

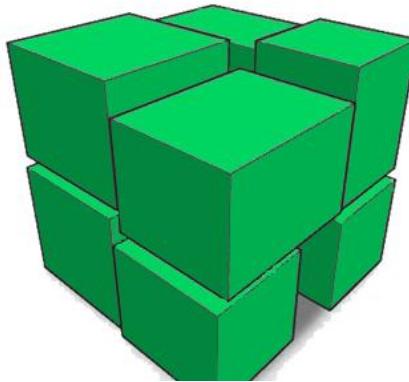
## The diversity of lime reagents



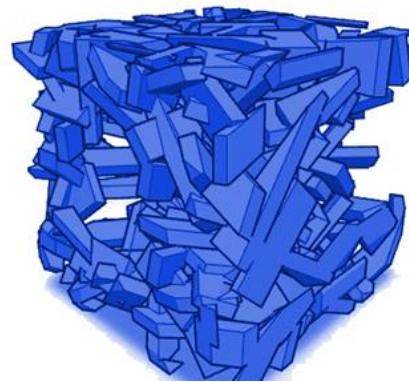
- Advanced Dry Sorbent Properties



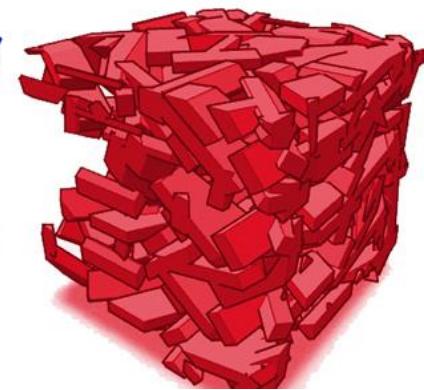
Local Hydrate



Sorbacal® H



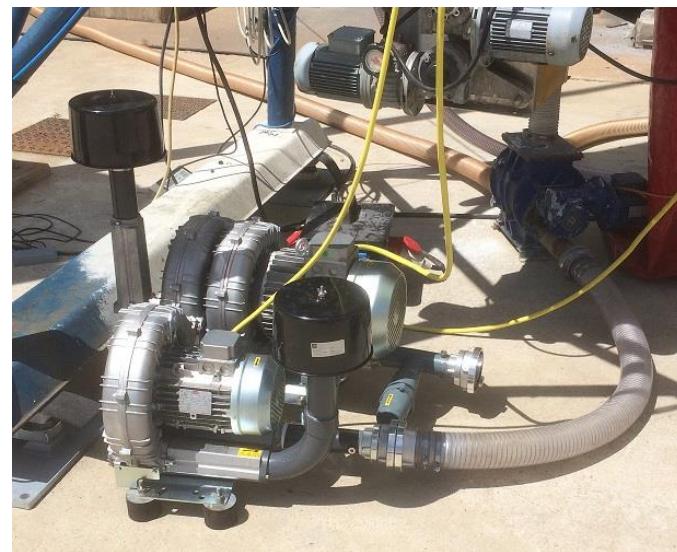
Sorbacal® SP



Sorbacal® SPS

Parameter	Local Hydrate	Sorbacal® H	Sorbacal® SP	Sorbacal® SPS	
Available Lime (EN-459)	70 - 90	>90	>90	>90	(wt%)
Surface Area (BET)	5 - 15	>20	>40	>40	(m <sup>2</sup> /g)
Pore Volume (BJH)	0.02 - 0.08	0.1	> 0.2	> 0.2	(ml/g)
Particle Size (d <sub>50</sub> )	4 - 25	5 - 14	5 - 14	5 - 14	(μm)
SO <sub>2</sub> removal performance	20 - 50	100	200	220 - 300	(%)

# Lime based dry sorbent injection at 20 MW pilot unit in CIUDEN

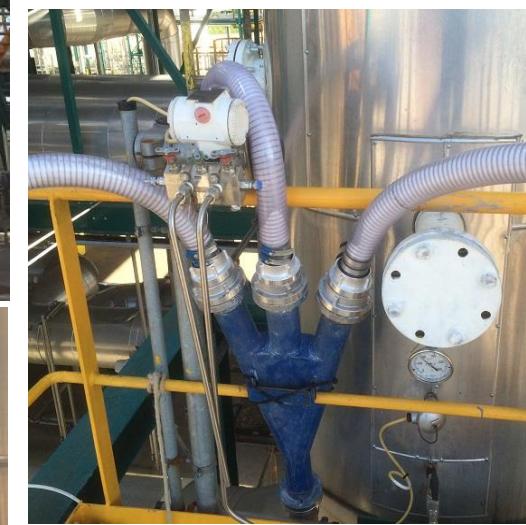


# Lime based dry sorbent injection at 20 MW pilot unit in CIUDEN

In furnace injection



Upstream baghouse  
injection

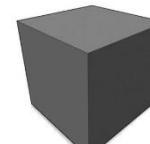


# Dry sorbents delivered to CIUDEN

- Sorbacal® SPS
  - High performance high porosity hydrated lime



- Finely milled natural limestone
- Activated carbon (35%) + milled limestone (65%)



Sample type	Finely milled limestone	Sorbacal® SPS	Activated Carbon GL 50 (Norit)
Humidity (%)	0,1	1,8	-
C total (%)	11,8	0,8	80,4
CaCO <sub>3</sub> (%)	98,3	6,4	-
Ca(OH) <sub>2</sub> (%)	0	93,3	-
Particle size distribution (Laser) - d50 (µm)	6,4	7,5	12,5
Particle size distribution (Laser) - d90 (µm)	24,2	39,3	88,7
Specific Surface Area BET (m <sup>2</sup> /g)	1,6	31,6	680
Total Porous Volume BJH (17-1000Å*) (cm <sup>3</sup> /g)	0,005	0,186	0,397
Partial Porous Volume BJH (100-300Å*) (cm <sup>3</sup> /g)	0,001	0,098	-
Bulk density (unpacked) (g/cm <sup>3</sup> )	0,83	0,49	0,26
Packed density (packed) (g/cm <sup>3</sup> )	1,22	0,61	0,38



# A little dose of chemistry..

## Hydrated lime



- $\text{Ca}(\text{OH})_2 + \text{SO}_2 \rightarrow \text{CaSO}_3 + \text{H}_2\text{O}$
- $\text{CaSO}_3 + \frac{1}{2}\text{O}_2 \rightarrow \text{CaSO}_4$
- $\text{Ca}(\text{OH})_2 + \text{SO}_3 \rightarrow \text{CaSO}_4 + \text{H}_2\text{O}$
- $\text{Ca}(\text{OH})_2 + \text{HCl} \rightarrow \text{CaClOH} + \text{H}_2\text{O}$
- $\text{CaClOH} + \text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{O}$

In furnace injection (above 900°C):

- $\text{Ca}(\text{OH})_2 \rightarrow \text{CaO} + \text{H}_2\text{O}$  **(very fast)**
- $\text{CaO} + \text{SO}_2 + \frac{1}{2}\text{O}_2 \rightarrow \text{CaSO}_4$
- $\text{CaO} + \text{SO}_3 \rightarrow \text{CaSO}_4$

But:  $\text{CaO} + \text{CO}_2 \rightarrow \text{CaCO}_3$  (below 850°C)



## Limestone



- ~~$\text{CaCO}_3 + \text{SO}_2 \rightarrow \text{CaSO}_3 + \text{CO}_2$~~
- ~~$\text{CaSO}_3 + \frac{1}{2}\text{O}_2 \rightarrow \text{CaSO}_4$~~
- ~~$\text{CaCO}_3 + \text{SO}_3 \rightarrow \text{CaSO}_4 + \text{CO}_2$~~
- ~~$\text{CaCO}_3 + 2 \text{HCl} \rightarrow \text{CaCl}_2 + \text{CO}_2$~~

Very slow,  
Limited to  
surface  
reaction  
(no porosity)

In furnace injection (above 900°C):

- $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  **(rather slow)**
- $\text{CaO} + \text{SO}_2 + \frac{1}{2}\text{O}_2 \rightarrow \text{CaSO}_4$
- $\text{CaO} + \text{SO}_3 \rightarrow \text{CaSO}_4$

But:  $\text{CaO} + \text{CO}_2 \rightarrow \text{CaCO}_3$  (below 850°C)



# SO<sub>2</sub> removal performance in CIUDEN

- Coal only : AIR mode vs. OXY mode

Fuel = 100% Coal

AIR mode vs. OXY mode

Sorbent = Sorbacal®SPS

Injection location: SCR bypass (350°C); baghouse filter (200°C)

**AIR mode flue gas composition :**  
 SO<sub>2</sub> = 550ppm; HCl = 3ppm; CO<sub>2</sub>=15%; H<sub>2</sub>O = 6%; O<sub>2</sub> = 3%

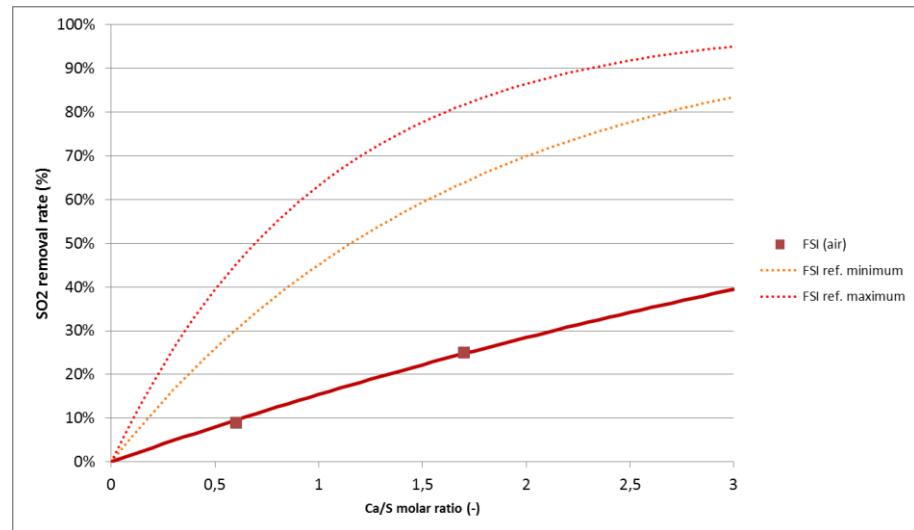
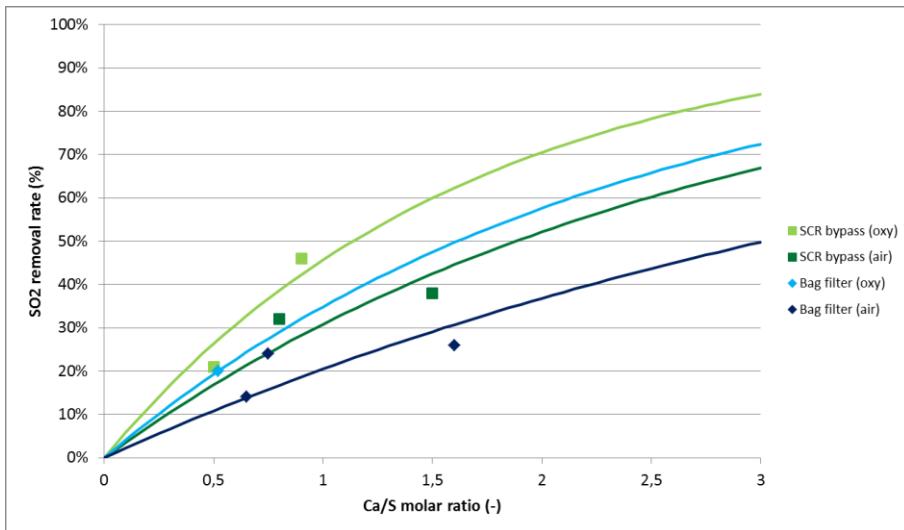
**OXY mode flue gas composition :**  
 SO<sub>2</sub> = 2550ppm; HCl = 5ppm; CO<sub>2</sub>=70%; H<sub>2</sub>O = 7%; O<sub>2</sub> = 3%

Fuel = 100% coal

AIR mode (OXY mode not available)

Sorbent = Sorbacal®SPS

Injection location: Furnace sorbent injection (FSI) (850°C)



- Order of magnitude in-line with expectations
- Better performance for **OXY** mode compared to Air mode
  - No detrimental effect of CO<sub>2</sub> despite high concentration (70%)
  - Due to 4-5x higher SO<sub>2</sub> concentrations

- FSI: performance significantly below expectations
  - Too low temperature, short residence time, lack of dispersion quality



# SO<sub>2</sub> removal performance in CIUDEN

- Coal + Biomass : AIR mode vs. OXY mode

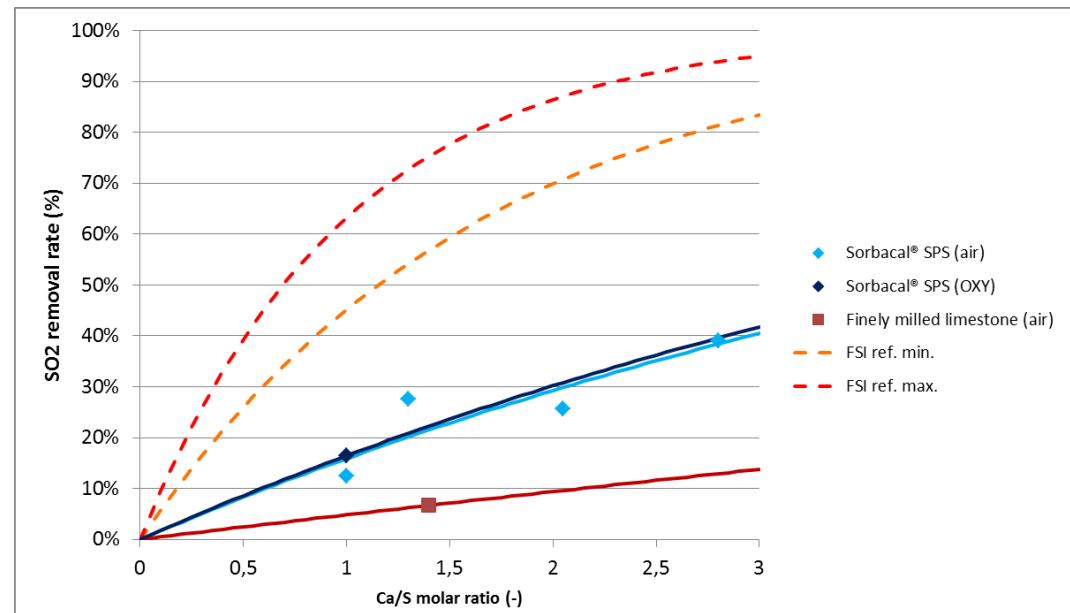
Fuel = Coal + Biomass (saw dust)

AIR mode vs. OXY mode

Sorbent = Sorbacal®SPS, finely milled limestone

Injection location: Furnace Sorbent Injection (FSI)

! Only FSI mode tested !



- Sorbacal® SPS :
  - performance significantly below expectations (Too low temperature, short residence time, lack of dispersion quality)
  - Similar performance in AIR and OXY mode
- Finely milled limestone: Much lower performance than Sorbacal® SPS in comparable test conditions
  - Much slower reaction with limestone

No suitable conclusion for SO<sub>3</sub>, HCl and Hg removal



# Fly ash analysis & valorization

- Valorization in cement: EN-450 standard

- Sulphate ( $\text{SO}_3$ ): < 3%
- Free CaO: < 1,5%
- Active CaO: < 10%
- Alcali ( $\text{Na}_2\text{O}$  eq.): < 5%
- Chlorides ( $\text{Cl}^-$ ): < 0,1%
- MgO: < 4%
- Reactive  $\text{SiO}_2$ : > 25%
- $\text{SiO}_2+\text{Al}_2\text{O}_3+\text{Fe}_2\text{O}_3$ : > 70%



- Other valorization ways: Soil stabilization, waste or sludge treatment & sanitization, ...
  - Site specific (chemical compatibility, legal aspects, commercial opportunities)
  - Much smaller market



# Fly ash analysis & valorization

## Fly ash analysis: Example

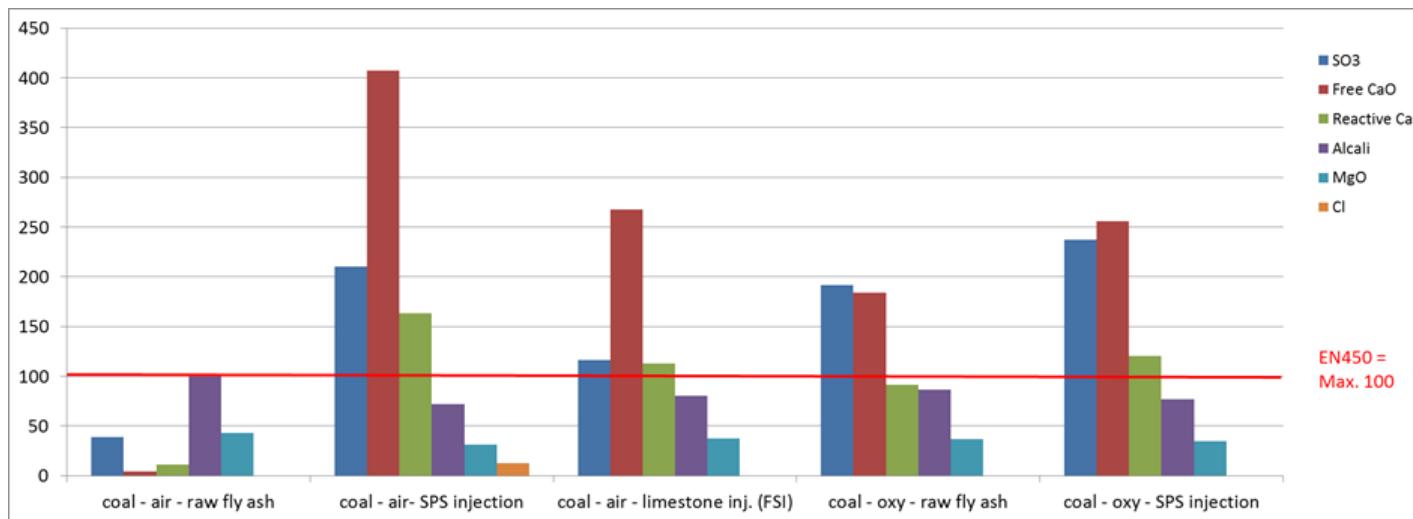
	Analysis Method	BE1116.269.2	BE1116.269.1	BE1116.269.3	BE1116.269.4	BE1116.269.6	BE1116.269.8	BE1116.269.7
<b>Client reference</b>		Biox S1-T1-A2	Biox S1-T1-A3	Biox S1-T1-A4	Biox S1-T2 -A2	Biox S1-T2 -A3	Biox S1-T3 -A2	Biox S1-T3 -A3
<b>Sample date</b>		22/05/2016	22/05/2016	23/05/2016	23/05/2016	23/05/2016	24/05/2016	24/05/2016
<b>sample description</b>		air-coal-raw fly ash-base case AIR mode	air-coal-raw fly ash-higher O2 excess	air-coal-raw fly ash, Air mode OFAS level 3	air-coal-furnace- SPS-low	air-coal-furnace- SPS-high	Air-coal-SCR- SPS-low	Air-coal-SCR- SPS-high
<b>DSI Sorbent type</b>		-	-	-	SPS	SPS	SPS	SPS
<b>Humidity (%)</b>	<b>LoI 150°C</b>	0,4	0,2	0,2	0,3	0,4	1,3	0,4
<b>Volumetric CO2 (%)</b>	<b>EN196-2</b>	0,0	0,4	0,0	0,0	2,0	1,8	1,7
<b>Available CaO (%)</b>	<b>EN196-2</b>	0,3	0,0	0,0	1,5	13,2	13,2	11,8
<b>eq. Ca(OH)2</b>	<b>calc.</b>	0,4	0,0	0,0	2,0	17,4	17,4	15,6
<b>Ca as CaO (%)</b>	<b>SQ XRF</b>	3,5	3,2	3,1	5,8	24,5	22,1	23,4
<b>Mg as MgO (%)</b>	<b>SQ XRF</b>	1,8	1,7	1,7	1,7	1,4	1,4	1,3
<b>S as SO3 (%)</b>	<b>SQ XRF</b>	1,3	1,3	1,0	1,5	6,3	9,0	7,5
<b>Cl (%)</b>	<b>SQ XRF</b>	0,0	0,0	0,0	0,0	0,0	0,1	0,0
<b>Na as Na2O (%)</b>	<b>SQ XRF</b>	2,8	2,8	2,9	2,8	2,1	2,1	2,2
<b>K as K2O (%)</b>	<b>SQ XRF</b>	3,5	3,2	3,1	3,1	2,3	2,3	2,3
<b>Si as SiO2 (%)</b>	<b>SQ XRF</b>	49,7	50,9	51,4	48,8	34,5	34,8	34,2
<b>Al as Al2O3 (%)</b>	<b>SQ XRF</b>	24,5	24,4	24,3	23,4	16,6	16,9	16,8
<b>Fe as Fe2O3 (%)</b>	<b>SQ XRF</b>	10,6	10,0	10,4	10,7	8,5	8,2	8,7
<b>P as P2O5 (%)</b>	<b>SQ XRF</b>	1,1	0,9	1,0	1,0	0,9	0,8	0,9
<b>Ti as TiO2 (%)</b>	<b>SQ XRF</b>	1,6	1,6	1,6	1,6	1,6	1,6	1,6
<b>Others (Mn, Sr, As, Cr, V, Zn...) (%)</b>	<b>SQ XRF</b>	0,3	0,3	0,3	0,3	0,3	0,2	0,3



# Fly ash analysis & valorization

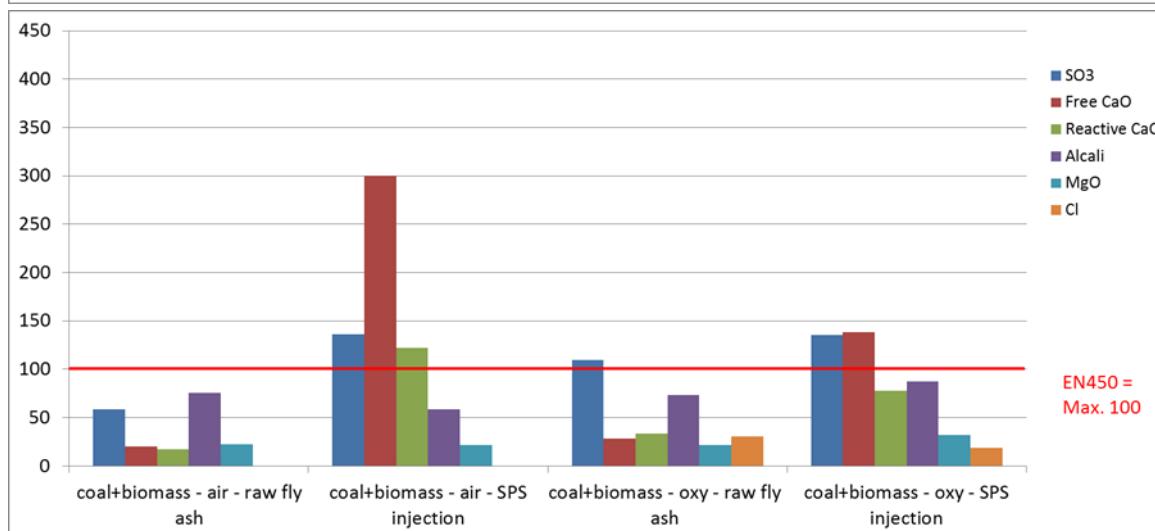
Valorization in cement: analysis vs. EN-450 standard

(BELOW 100 for compliance)



Coal  
only

EN450 =  
Max. 100



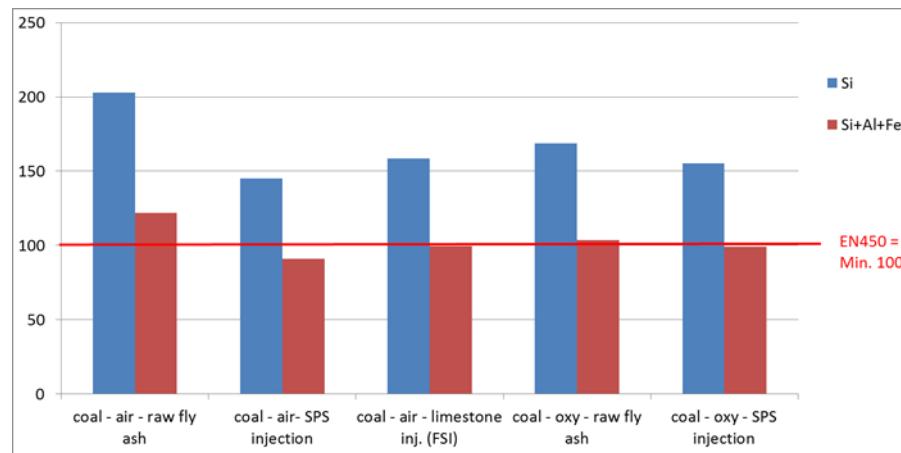
Coal +  
Biomass



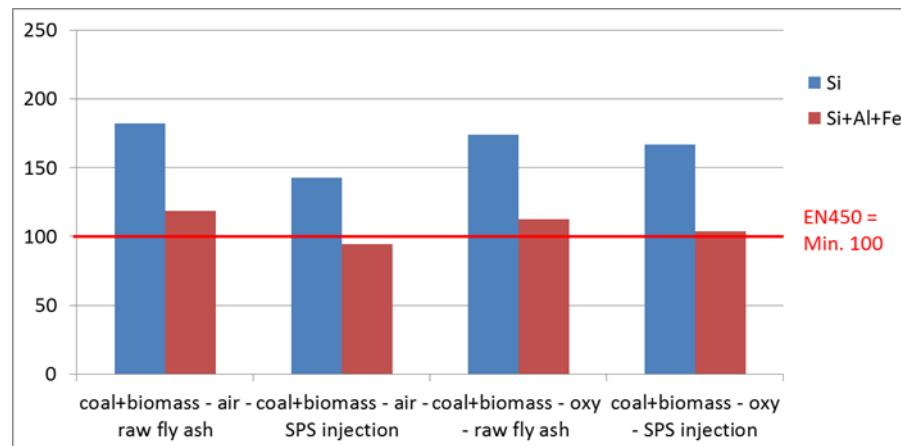
# Fly ash analysis & valorization

Valorization in cement: analysis vs. EN-450 standard

(ABOVE 100 for compliance)



Coal  
only

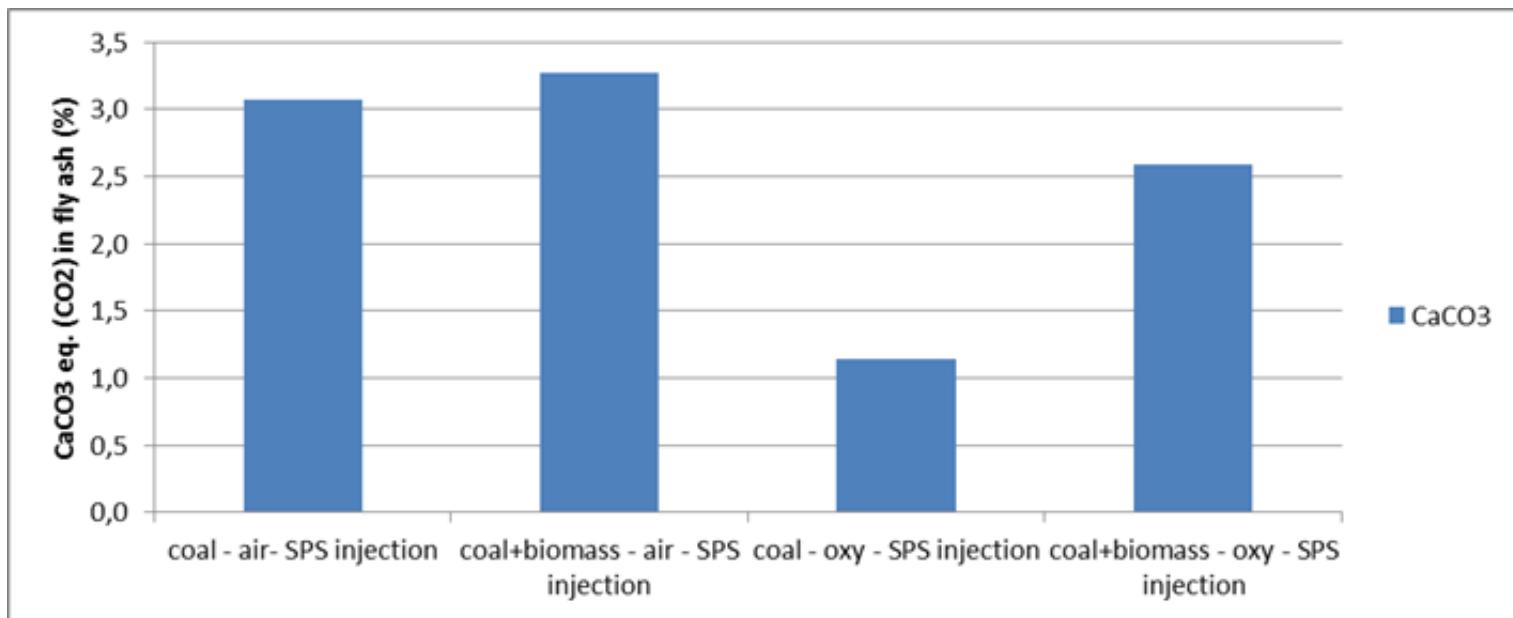


Coal +  
Biomass



# Impact of OXY mode on sorbent carbonation

$\text{CaCO}_3$  content in fly ash (with sorbent injection)



**NO additional detrimental effect of  $\text{CO}_2$  in OXY mode compared to AIR mode**

( $\text{CO}_2$  concentration = 68-73% in OXY mode)



## Conclusions

- Successful delivery and operation of the mobile sorbent dosing units + sorbents in large scale (20 MW) pilot test facility in CIUDEN
- SO<sub>2</sub> removal performance :
  - Performance fair agreement with expectation for 350°C injection (SCR bypass) and 200°C injection (upstream baghouse filter)
  - Much lower performance than expected in Furnace Sorbent Injection (850°C), due to inadequate temperature & residence time.
- Higher SO<sub>2</sub> removal performance in OXY mode compared to AIR mode
  - **No additional detrimental effect of CO<sub>2</sub> in OXY mode (CO<sub>2</sub> > 68%)**
  - SO<sub>2</sub> concentration 4-5x higher in OXY mode
- No suitable conclusion for SO<sub>3</sub> & HCl removal
  - too low concentrations
- No suitable conclusion for Hg removal by activated carbon
  - Too short trial, no filter vaccination



# Conclusions

- Fly ash valorization:
  - Fly ash without acidic gas sorbent contamination
    - AIR mode: Fly ash are compliant for valorization in cement (EN 450)
    - OXY mode: Fly ash not always compliant with EN 450 ( $\text{SO}_3$  content)
  - Fly ash with lime-based acidic gas sorbent contamination
    - In the conditions of the tests, fly ash not compliant with EN 450 standard, due to following criteria:
      - ✓ Most of the time:  $\text{SO}_3$  content, Free CaO.
      - ✓ Sometimes:  $\text{Reactive CaO; SiO}_2+\text{Al}_2\text{O}_3+\text{Fe}_2\text{O}_3$
  - Fly ash with alcali-based acidic gas sorbent contamination
    - No compliance with EN 450 standards, due to Alcali ( $\text{Na}_2\text{O}$  eq.) and  $\text{SO}_3$  content
    - Alkali are highly detrimental to cement applications and strictly regulated
- Potential valorization of fly ash contaminated with lime-based by-product for applications with less stringent constraints, depending on local contingencies
  - Soil stabilization & hardening
  - Waste or sludge stabilization & sanitization
  - Neutralization of acidic streams, ...

